

MODULE SPECIFICATION

| Part 1: Information | | | | | | | |
|---------------------------|----------------|-----------------------------------|--------------------|---------------------------|--|--|--|
| Module Title | Comb | bined Sciences | | | | | |
| Module Code | UZYREU-15-0 | | Level | Level 3 | | | |
| For implementation from | 2020- | -21 | | | | | |
| UWE Credit Rating | 15 | | ECTS Credit Rating | 7.5 | | | |
| Faculty | Facul Scien | ty of Health & Applied ces | Field | Allied Health Professions | | | |
| Department | HAS | Dept of Allied Health Professions | | | | | |
| Module Type: | Stanc | Jard | | | | | |
| Pre-requisites N | | None | | | | | |
| Excluded Combinations | | None | | | | | |
| Co-requisites | | None | | | | | |
| Module Entry Requirements | | None | | | | | |
| PSRB Requirements | | None | | | | | |

Part 2: Description

Educational Aims: See Learning Outcomes.

Outline Syllabus: Basic Concepts SI units, including multiples and sub-multiples Measuring and recording of quantities using standard laboratory equipment Interpreting experimental results Producing a laboratory report Using and manipulating relevant physical formulae Mass, weight and force

Energy and Energy Transfers Energy, work, power and efficiency Heat energy, specific heat capacity and heat transfer Kinetic and potential energy

Waves Wave properties and wave equation ($\Box = \Box \Box$)

Refraction, reflection, endoscopy Atoms and Bonding Atomic structure (Bohr model of atom) Atomic number, mass number and isotopes Periodic Table Elements and compounds Covalent, ionic bonding Simple chemical equations Solubility Radioactivity Types of radiation, \Box , \Box , including properties and production Radioactive decay, half-life and nuclear equations

Electricity Current, voltage (potential difference). Generation, resistance and power Simple circuits including series and parallel circuits Circuit symbols and circuit diagrams Ohm's Law

Forces Scalars and vectors Adding and resolving vectors Force-extension relationships (Hooke's Law) for springs and some materials Elastic and plastic behaviour Turning effect of forces (moment/torque) Equilibrium

Dynamics Distance, displacement, speed, and velocity Newton's laws of motion

Acidity Acids, bases and alkalis H+ ions, pH scale Neutralisation Acidosis & alkalosis

Chemical Kinetics Reaction rates, factors affecting rates Simple collision theory Catalysis

Teaching and Learning Methods: This module operates on the basis of 150 hours of study in total.

This includes 90 hours of scheduled teaching (comprising 35 hours of lectures, 10 hours of tutorials, 35 hours of laboratory sessions, and 10 hours of project supervision) plus 60 hours of independent study.

Scheduled learning may include a combination of face to face and online lectures, small group activities, supervised practicals, and demonstration.

Independent learning includes hours engaged with essential reading, assignment preparation and completion etc. These sessions constitute an average time per level as indicated in the table below.

Part 3: Assessment

Regular formative assessment will take place throughout the module delivery to enable students to gauge their progress and learning to date.

Component A is a 1.5 hour exam covering physics and chemistry. This component offers the opportunity for students to demonstrate calculation skills.

Component B is a Laboratory Report (1,250 words). This allows the student to demonstrate skills in presenting the outcome of a laboratory based exercise, in an appropriate and professional manner.

| First Sit Components | Final Assessment | Element weighting | Description |
|------------------------------------|---------------------|----------------------|---|
| Laboratory Report - Component B | ✓ | 30 % | Laboratory Report (1,250 words) |
| Examination - Component A | | 70 % | Exam (1.5 hour) (Physics and Chemistry) |
| Resit Components | Final Assessment | Element weighting | Description |
| Examination - Component A | | 70 % | Exam (1.5 hour) (Physics and Chemistry) |
| Laboratory Report - Component B | ~ | 30 % | Laboratory Report (1,250 words) |

| Part 4: Teaching and Learning Methods | | | | | |
|---|--|---|--|--|--|
| On successful completion of this module students will achieve the follo | wing learning | outcomes: | | | |
| Module Learning Outcomes | | | | | |
| Demonstrate a knowledge of the SI system of units | MO1 | | | | |
| Demonstrate a knowledge of atomic structure, and chemical bonding | | | | | |
| Understand the simple wave model, electricity, radiation and radioact | MO3 | | | | |
| Explain and apply the laws of motion and the concepts of forces including weight | | | | | |
| Understand simple chemical kinetics and demonstrate knowledge of acids and alkalis | | | | | |
| Demonstrate analytical ability in a range of situations and evaluate material appropriately | | | | | |
| Apply theoretical knowledge to practical contexts | | MO7 | | | |
| Formulate and test hypotheses, record data and draw appropriate conclusions | | | | | |
| Perform standard laboratory procedures | MO9 | | | | |
| Establish and record observations and experimental work | | | | | |
| Independent Study Hours: | | | | | |
| Independent study/self-guided study | 6 | 60 | | | |
| Total Independent Study Hours: | 6 | 0 | | | |
| | Part 4: Teaching and Learning Methods On successful completion of this module students will achieve the follo Module Learning Outcomes | Part 4: Teaching and Learning Methods On successful completion of this module students will achieve the following learning Module Learning Outcomes Demonstrate a knowledge of the SI system of units Demonstrate a knowledge of atomic structure, and chemical bonding Understand the simple wave model, electricity, radiation and radioactive decay Explain and apply the laws of motion and the concepts of forces including weight Understand simple chemical kinetics and demonstrate knowledge of acids and alkalis Demonstrate analytical ability in a range of situations and evaluate material appropriately Apply theoretical knowledge to practical contexts Formulate and test hypotheses, record data and draw appropriate conclusions Perform standard laboratory procedures Establish and record observations and experimental work Independent Study Hours: Independent study/self-guided study 6 | | | |

| | Scheduled Learning and Teaching Hours: | |
|-----------------|--|-----|
| | | |
| | Face-to-face learning | 90 |
| | Total Scheduled Learning and Teaching Hours: | 90 |
| | | |
| | | |
| | Hours to be allocated | 150 |
| | Allocated Hours | 150 |
| Reading List | The reading list for this module can be accessed via the following link: | |
| | https://uwe.rl.talis.com/index.html | |

This module contributes towards the following programmes of study: